Roll No.

Total Pages: 3

GSE/M-22

1481

ORGANIC CHEMISTRY Paper-CH-106

Time: Three Hours]

[Maximum Marks: 32

Note: Attempt five questions in all. Question No. 1 is compulsory. Select two questions from each Section.

Compulsory Question

1. (a) Give IUPAC names of the following compounds:

- (b) Out of cis 2-Butene and trans 2-Butene, which has higher boiling point, and why?
- (c) Give the name and structure of electrophiles generated in Nitration and Friedel Craft Alkylation reactions in aromatic electrophilic substitution.
- (d) Give the preparation of Ethyl bromide by Hunsdiecker reaction. (2×4=8)

SECTION-A

 (a) Explain Saytzeff rule with the help of dehydrohalogenation of 2-Bromobutane with alc. KOH. (b) Explain in detail the rearrangement process occurring during the addition of HBr to 3-Methylbut-l-ene.

(3,3)

- (a) Write the reaction and mechanism of dehydration of 2-Butanol with conc. H₂SO₄ to give alkene.
 - (b) Discuss the reaction and mechanism of oxidation of 2-Methylbut-2-ene with perbenzoic acid. Also give the ring opening reaction of the resulting epoxide in basic medium. (3,3)
- 4. (a) Explain o, p-directing and activating nature of -NH₂ group.
 - (b) What is meant by Aromaticity? State Hückel Rule and explain with suitable examples. (3,3)
- 5. (a) What are Annulenes? Give one example each of an aromatic, antiaromatic and non-aromatic annulene.
 - (b) Give the reaction and mechanism of Sulphonation of benzene. (3,3)

SECTION-B

- 6. (a) What are Dienes? Give their classification with one rexample each.
 - (b) Explain the acidic nature of terminal alkynes. (3,3)
- (a) Explain the addition of HBr to 1,3-Butadiene along with mechanism.
 - (b) Write the reaction and mechanism of Birch reduction of 2-Butyne with Na-Liq.NH₃. (3,3)

[P.T.O.

- 8. (a) Give Addition-Elimination mechanism of Nucleophilic Aromatic Substitution in Aryl halides.
 - (b) Convert CH₃Br into (i) CH₃NH₂ (ii) CH₃CN (iii) CH₃OCH₃. (3,3)
 - (a) Discuss the mechanism and stereochemistry of S_{N1} reactions.
 - (3,3)

 One of the factors affecting S_{N2} reactions. (3,3)

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